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Relationship between plant uptake of pesticides and water-extractable residue in Japanese soils

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The relationship between pesticide concentrations in Komatsuna (Brassica rapa var. perviridis) shoots and the extractable concentrations in soils by a sequential soil-extraction method was investigated in 8 pesticides and in 4 soils. Concentrations of many pesticides in Komatsuna shoots showed higher positive correlation with water-extractable than with total-extractable soil concentrations. We also examined the effects of the soil-aging periods (SAPs) between pesticide applications and sowing on pesticide concentrations in Komatsuna shoots. Pesticide concentrations in Komatsuna shoots and soil-water extracts decreased with increasing SAPs. These results imply that estimating the pesticide concentrations in crops based on water-extractable concentrations in soil and setting the appropriate SAPs prevent contamination of crops by pesticide residues in soil. However, the correlation between pesticide concentrations in Komatsuna shoots and water extracts in soils, and the effectiveness of pesticide reduction in Komatsuna shoots by setting long SAPs differed according to the pesticide type. Hence, the applicability of the water-extraction method should be verified for each pesticide. © Pesticide Science Society of Japan

Keywords: pesticides, plant uptake, succeeding crops, water-extraction of soil, phytoavailability, soil aging.

Electronic supplementary materials: The online version of this article contains supplementary materials (Supplemental Figures S1–S7, Supplemental Tables S1–S11), which are available at http://www.jstage.jst.go.jp/browse/jpestics/.

Introduction

In Japan, some agricultural chemicals were recently detected in crops that exceeded the uniform residue limit of 0.01 ppm set by the Japanese Positive List System.¹⁾ One reason for this is that succeeding crops take up from the soil residual pesticides applied to previous crops. Under the Japanese system of registering newly developed pesticides, the studies for pesticide residue in succeeding crops are required only if the half-lives of pesticides in the soil are longer than 100 days²⁾; therefore, in many cases, residue studies for succeeding crops are not conducted. In contrast, in the United States, confined accumulation studies for rotational crops are required as a first tier evaluation for all pesticides except for those applied only on semi-permanent or permanent crops³⁾; these results are available for setting appropriate plant-back intervals (PBIs), *i.e.*, plant restriction intervals, which are included on the pesticide label.

* To whom correspondence should be addressed. E-mail: ymotoki@affrc.go.jp Published online October 28, 2015 © Pesticide Science Society of Japan Recent studies of crops contaminated by persistent organic pollutants (POPs), including dieldrin and heptachlor *exo*-epoxide, showed that application of carbonaceous adsorbents, such as activated carbon, to soils⁴⁻⁶⁾ and a grafting technique using a low-uptake rootstock^{4,7)} effectively decreased concentrations of POPs in cucurbit crops. However, the pesticides on the market are more degradable than POPs in soils. Therefore, estimating the residue concentrations in crops based on those in soils, followed by setting appropriate PBIs, is considered as one of realistic measures for preventing the production and circulation of contaminated crops because there is no cost for the materials. Understanding the dissipation behavior of pesticides in soils and the relationship between the residual levels of pesticides in soils and those in crops is necessary for setting appropriate PBIs.

Residual levels of pesticides in soils are generally determined by using a vigorous extraction technique of soils with an organic solvent, such as acetone. However, assuming that some pesticides sorbed to soil particles are eluted into soil water and, subsequently, taken up by plant roots, a vigorous extraction technique that extracts not only water-extractable pesticides but also soil-sorbed pesticides is inadequate for assessing the amount of phytoavailable pesticides in soils. In fact, Sugimura *et al.*^{8,9)}

found that the phytotoxicity of pendimethalin and metolachlor in upland soils increased with increasing the concentration of ingredients extracted from soils by distilled water. Although they did not provide the residual concentration of the ingredients in plants, their results imply that quantifying the amount of water-extractable pesticides in soils is important for predicting the amount of plant uptake from soils. The water extractability of pesticides in soils depends on soil sorption. It is generally known that the main sorptive medium of nonionic pesticides in soil is organic matter, i.e., neutral organic chemicals, including pesticides that sorb to soil organic matter via hydrophobic interactions. 10) Furthermore, it has been reported that the sorption of a given pesticide shows high variability among Japanese soils because of andosol, the typical upland soil in Japan, which contains high organic carbon (OC) content.¹¹⁾ Therefore, the phytoavailability of pesticides in Japanese soils is also likely to differ considerably.

The objectives of this study are to investigate the variability of pesticide concentrations in Komatsuna (*Brassica rapa* var. *perviridis*), as a representative of crops, cultivated in four typical Japanese soils, to assess the relationships between pesticide concentrations in Komatsuna and water-extractable concentrations in soils, and to elucidate the effect of the soil-aging periods (SAPs) between applications of pesticides and sowing on residue

concentrations in Komatsuna.

Materials and Methods

1. Pesticides

The eight pesticides used in this study (chemical purity >97.5%) were purchased from Wako Pure Chemical Industries, Ltd. (Osaka, Japan), Kanto Chemicals (Tokyo, Japan), and Dr. Ehrenstorfer GmbH (Augsburg, Germany) (Table 1). The pesticides were chosen based on their hydrophobicity, *i.e.*, the range of their octanol-water partition coefficients ($\log K_{\rm ow}$). To enable simultaneous analysis, the pesticides were divided into two analytical groups.

2. Plant uptake experiments

Four typical Japanese agricultural soils were selected to cover a wide range of OC contents (Table 2). Soil samples were airdried, passed through a 2.0-mm sieve, and used in pot experiments. To evenly mix pesticides with soils, a mixture of Celite powder and pesticides was prepared. Approximately 1.5 L of acetone solution containing 80 mg of each pesticide was added to 400 g of Celite 545 (Wako Pure Chemical Industries, Ltd.), and the acetone was evaporated for 5 hr at room temperature in a draft chamber. The concentrations of pesticides in the Celite powder were quantified by acetone extraction; they ranged from

Table 1. Octanol-water partition coefficient ($\log K_{ow}$) and analytical groups of test compounds

Compound	IUPAC name	Substance group	$\log K_{\mathrm{ow}}^{}(a)}$	$\log K_{\mathrm{ow}}^{}}$	$\log K_{\mathrm{ow}}^{\mathrm{c})}$	Analytical group ^{d)}
Imidacloprid	(<i>E</i>)-1-(6-chloro-3-pyridylmethyl- <i>N</i> -nitroimidazoli-din-2-ylideneamine	Neonicotinoid	0.57	0.57	0.57	A
Clothianidin	(<i>E</i>)-1-(2-chloro-1,3-thiazol-5-ylmethyl)-3-methyl-2-nitroguanidine	Neonicotinoid	0.91	0.70	0.70	A
Thiacloprid	(<i>Z</i>)-3-(6-chloro-3-pyridylmethyl)-1,3-thiazolidin-2-ylidenecyanamide	Neonicotinoid	1.26	0.74	1.26	A
Metalaxyl	Methyl N -(methoxyacetyl)- N -(2,6-xylyl)-DL-alaninate	Phenylamide	1.65	1.75	1.75	A
Fosthiazate	(RS)-S-sec-butyl O-ethyl 2-oxo-1,3-thiazolidin- 3-ylphosphonothioate	Organophosphate	1.68	1.68	1.68	A
Flutolanil	α, α, α -trifluoro-3'-isopropoxy- o -toluanilide	Phenylbenzamide	3.17	3.17	3.77	В
Procymidone	<i>N</i> -(3,5-dichlorophenyl)-1,2-dimethylcyclopropane-1,2-dicarboximide	Dicarboximide	3.30	3.14	3.30	В
Tetraconazole	$(RS)\hbox{-}2\hbox{-}(2,4\hbox{-}dichlorophenyl)\hbox{-}3\hbox{-}(1H\hbox{-}1,2,4\hbox{-}triazol\hbox{-}1\hbox{-}yl)\\propyl 1,1,2,2\hbox{-}tetrafluoroethyl ether}$	Triazole	3.56	3.56	3.53	В

 $^{^{}a)}\log K_{\mathrm{ow}}$ values were obtained from the Footprint Pesticide Properties Database of IUPAC. $^{12)}$ $^{b)}\log K_{\mathrm{ow}}$ values were obtained from The Pesticide Manual (16th ed.). $^{(13)}$ $^{c)}\log K_{\mathrm{ow}}$ values were obtained from The 2011 Pesticide Handbook. $^{(14)}$ Analytical methods are described in Supplemental Figs. S1–S3.

Table 2. Properties of test soils^{a)}

Soil	Classification ^{b)}	Texture ^{c)}	OC ^{d)} (%)	CEC ^{e)} (cmol(+)/kg)	Clay (%)	pH (H ₂ O)	ECf) (mS/cm)
PS1	Yellow soil	Light clay	1.02	11.4	39.0	5.3	0.09
PS2	Gray lowland soil	Silty clay	1.46	18.2	25.3	5.8	0.17
PS3	Andosol	Loam	5.21	33.8	10.8	5.5	0.15
PS4	Andosol	Silty loam	8.65	35.4	1.8	5.8	0.05

^{a)} Data were obtained from our previous report.¹¹⁾ Soil samples PS1, PS2, PS3, and P4 correspond to Soils S3, S5, S7, and S8 in the previous report, respectively. ^{b)} According to the criteria adopted by the Cultivated Soil Classification Committee.^{15) c)} According to the standards developed by the International Society of Soil Science. ^{16) d)} Organic carbon content. ^{e)} Cation exchange capacity. ^{f)} Electrical conductivity.

84 to $137 \mu g/g$. Each air-dried soil sample (PS1, 630 g; PS2, 550 g; PS3, 450 g; PS4, 450 g), in the amount one hundred times greater than the amount of the Celite powder, was spread in a plastic bag; then Celite powder, 2g of chemical fertilizer [0.16g of N as $(NH_4)_2SO_4$, 0.07 g of P as $Ca(H_2PO_4)_2 \cdot H_2O$, and 0.14 g of K as KCl], and 2g of dolomite were evenly sprinkled onto the soils and mixed well. The theoretical concentrations ($\mu g/g$) of pesticides in air-dried soils were as follows: imidacloprid, 1.31; clothianidin, 1.33; thiacloprid, 1.37; metalaxyl, 1.24; fosthiazate, 1.33; flutolanil, 1.06; procymidone, 0.84; tetraconazole, 1.10.

Plastic pots (600 mL) were filled with the soil samples. To improve the germination rate, approximately 5g of perlite was sprinkled onto the soil surface and lightly mixed. Each pot was set in a plastic tray and placed in a climate chamber. The seeds of Komatsuna (B. rapa var. perviridis 'Yokattana') were sowed in each pot after 7 days for soils PS1 and PS4 and after 7, 35, and 63 days for soils PS2 and PS3. Thus, an SAP of 7 days was set for all soils as a base, and SAPs of 35 and 63 days were additionally set for soils PS2 (gray lowland soil) and PS3 (andosol). Komatsuna was chosen because it is a popular leafy vegetable in Japan, is easy to cultivate, and grows rapidly. The water content of the soils in the pots was adjusted by bottom-up irrigation to between approximately 50% and 70% of their water holding capacity (WHC), except for on sowing day (90% of WHC). The climate chamber was set to a cycle of 13 hr of light and 11 hr of darkness at 25°C, except during the first 7 days after the application of Celite to the soil and the period between sowing and germination (24hr of darkness at 25°C). During the later growing period, i.e., the 18-day period until harvest, 50 mL of liquid fertilizer (at two-fold the strength of Otsuka Chemical A prescription [mg/L]: N, 520; P, 104; K, 672; Ca, 328; Mg, 72; Mn, 2.4; B, 1.0; Fe, 5.4; Cu, 0.06; Zn, 0.18; Mo, 0.06) was applied 10 times to each pot by bottom-up irrigation. Twenty-eight days after sowing, the shoots, including the leaves and stem, were harvested. The fresh weight (FW) of the shoots ranged from 59.7-87.8 g per pot. Pot experiments were performed in quadruplicate.

The shoot samples were frozen and finely ground using a Grindomix GM200 (Retsch GmbH, Haan, Germany). Ten grams of aliquot was soaked in 80 mL acetone for 30 min and homogenized using a Polytron® PT3100 (Kinematica AG, Lucerne, Switzerland) for 5 min. The solution was filtered through a 0.8-µm glass fiber filter and the volume was made up to 100 mL with acetone.

3. Soil dissipation experiments

Soil dissipation experiments were performed in triplicate under the same conditions as the plant uptake experiments except that planting of Komatsuna and application of perlite and additional liquid fertilizer were not carried out. Approximately 20 g of soil was taken from each pot at 7, 14, 21, and 35 days for soils PS1 and PS4, and at 7, 14, 21, 35, 63, and 91 days for soils PS2 and PS3 after Celite was applied to the soil. Each soil sample was divided into two subsamples: one was dried at 110°C to measure the moisture content, and the other was analyzed using a sequential extraction method.

4. Sequential extraction of soils

Water extraction of the soils was carried out in accordance with the standard batch sorption test method, 19) although distilled water rather than a solution of calcium chloride was used as the aqueous solvent. Twenty-five milliliters of distilled water was added to approximately 5 g dry weight (DW) of soil in a 50-mL glass centrifuge tube. The tubes were agitated on a thermostat shaker (TAITEC, Saitama, Japan) for 24 hr at 25±2°C in the dark. After shaking, the mixtures were centrifuged at $1200 \times g$ for 30 min. Fifteen milliliters of supernatant was taken and used to quantify the water extracts. In addition, 30 mL of acetone was added to the remaining sample, the tubes were shaken in a thermostat shaker for 20 min at $25\pm2^{\circ}$ C and centrifuged at $1200\times g$ for 10 min, and the supernatant was carefully decanted. This extraction procedure was repeated twice. The collected supernatant was evaporated in a rotary evaporator to reduce the volume to approximately 10 mL and then used to quantify the acetone extracts.

The mass fraction of water-extractable pesticides in the soil, C_{water} (μ g/g DW), was calculated using the following equation:

$$C_{\text{water}} = C_{\text{aq}} \cdot (V_{\text{add}} + V_{\text{sw}}) / M, \qquad (1)$$

where $C_{\rm aq}$ ($\mu g/mL$) is the mass concentration of pesticide in the aqueous phase after shaking for 24hr, Vadd is the volume of added distilled water (mL), V_{sw} is the volume of soil water (mL), and M is the soil's dry mass (g). The mass fraction of total extractable pesticides in the soil, C_{total} (μ g/g DW), was represented by the following equation:

$$C_{\text{total}} = \left(C_{\text{aq}} \cdot V_{\text{aliq}} + m_{\text{E}}\right) / M, \qquad (2)$$

where $V_{\rm aliq}$ is the volume of the aliquot taken from the supernatant (mL) and m_E is the mass of the pesticides extracted by acetone from the remaining sample after V_{aliq} was removed (μ g).

The half-lives $(\tau_{1/2})$ of C_{water} and C_{total} were calculated using the following equations 18,20):

$$k = -[\ln(C_t / C_0)]/t \tag{3}$$

$$\tau_{1/2} = -\ln(2)/k$$
, (4)

where k is the first-order rate constant (1/day), C_t is the C_{water} or C_{total} after t days, and C_0 is the initial C_{water} or C_{total} .

The mean C_{water} or C_{total} values during the periods from t_1 to t_2 were calculated by integrating equation (3) from t_1 to t_2 and dividing the integration values by the periods $(t_1 \text{ to } t_2)$. ¹⁸⁾

5. Pesticide analysis

Acetone extracts from Komatsuna shoots were analyzed to quantify the pesticides using two different methods, as shown in Supplemental Figs. S1 and S2. For group A, 10 mL of aliquot was taken from each acetone extract and evaporated in a rotary evaporator; 10 mL of acetonitrile was then added to each concentrate. The mixtures were cleaned with an ENVI-Carb II/PSA

column (500 mg/500 mg; Supelco, Bellefonte, USA). For group B, a 50-mL aliquot of each acetone extract was evaporated and cleaned with a diatomite column (InertSep K-solute 20 mL; GL Sciences, Tokyo, Japan) followed by an ENVI-Carb II/PSA column (500 mg/500 mg; Supelco).

The aliquots of water extracts and concentrates of acetone extracts from soils were also analyzed using two different methods, 11) whereas group B was analyzed using the method applied for the analysis of Komatsuna shoots (Supplemental Fig. S2). The samples from group A were cleaned with a diatomite column (InertSep K-solute 20 mL; GL Sciences) followed by a PSA column (500 mg; Supelco) (Supplemental Fig. S3).

Cleaned samples were analyzed by liquid chromatographytandem mass spectrometry (LC-MS/MS) for group A (Supplemental Tables S1 and S2) and gas chromatography-mass spectrometry (GC-MS) for group B (Supplemental Tables S3 and S4). In the case of the high matrix effects of purified samples, matrix-matched standards were used in LC-MS/MS and GC-MS.²¹⁾

Pesticide recovery tests were performed on pulverized samples of Komatsuna shoots, distilled water, and test soils. Komatsuna shoots (10 g) and distilled water (15 mL) spiked at 3 and 1 ng/g, respectively, for all groups of pesticides. However, each soil (5g) spiked at 5 and 3 ng/g for group A and group B pesticides, respectively. Those samples were analyzed using the above-mentioned methods; however, 10 mL of distilled water was added to soil samples prior to acetone extraction. The mean recovery from five replicates for all compounds, except distilled water and soils for group A (four replicates), ranged from 73.3-113.0%; the coefficients of variation (CVs) were below 14.4% (Supplemental Tables S5-S7). The limits of quantification (LOQs) for pesticide analysis were calculated in accordance with Japanese Industrial Standard (JIS) K 0312.²²⁾ The LOQs for all compounds in Komatsuna shoots, distilled water, and all soil samples ranged from 0.94-2.09 ng/g, 0.16-1.29 ng/mL, and 0.57-3.16 ng/g, respectively (Supplemental Tables S5-S7).

6. Statistical analysis

All statistical analyses, including linear regression analysis and analysis of variance (ANOVA), were performed using SPSS Statistics 19 (IBM Corporation, Armonk, USA).

Results and Discussion

 Relationships between pesticide concentrations in Komatsuna shoots and water- or total-extractable concentrations in soils

Fig. 1 shows the pesticide concentrations in Komatsuna shoots cultivated in four soils with a 7-day SAP. The residue concentrations in Komatsuna shoots for a given pesticide varied among test soils, and CVs of the concentrations for a given pesticide ranged from 17.6% for metalaxyl to 123.5% for thiacloprid. Pesticide concentrations in Komatsuna shoots given the residue concentration in soils were analyzed using linear regression analysis between pesticide concentrations in Komatsuna shoots cultivated in soils with a 7-day SAP and the mean C_{water} or C_{total} values during the 28-day period until harvest, i.e., the period between sowing and harvest, in four soils (Table 3 and Supplemental Figs. S4 and S5). The regression lines were calculated using data containing the point (0, 0) because if the concentration in soil is zero, the concentration in Komatsuna shoot is also zero. Although the coefficient of determination (R2) values differed according to the type of pesticide, the concentrations of pesticides, except for metalaxyl and fosthiazate, in Komatsuna shoots were more strongly correlated with the C_{water} than with the C_{total} .

This result suggests that pesticide concentrations in Komatsuna shoots were affected more by the $C_{\rm water}$ than by the $C_{\rm total}$, which implies that the water-extraction method of soil is more suitable than the total-extraction method for estimating residual concentrations in crops. On the other hand, for metalaxyl and fosthiazate, the R^2 values were also high when using the $C_{\rm total}$. The variability of metalaxyl and fosthiazate concentrations in Komatsuna shoots was especially low; the CVs of these pesticides were less than 30%. Therefore, for these two pesticides, the regression lines were calculated by a two-point data group, the point (0,0) and the others (Supplemental Fig. S5). Focusing on imidacloprid, clothianidin, flutolanil, and tetraconazole,

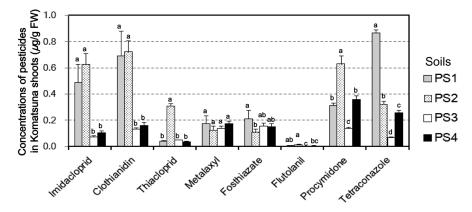


Fig. 1. Pesticide concentrations in Komatsuna shoots grown in soils with a 7-day soil-aging period. Columns with the same letter are not significantly different at p < 0.05, according to ANOVA-protected Tukey's multiple range testing. Error bars indicate standard deviations (n = 4).

Table 3. Linear regression analysis^{a)} between pesticide concentrations in Komatsuna shoots cultivated on soils with a 7-day SAP^{b)} and extractable concentrations in four soils

	No. soils	R^2						
Compound		Water-extracts			Total-extracts			
		Mean conc. during 28-day until harvest ^{c)}	Conc. at sowing	Conc. at harvest	Mean conc. during 28-day until harvest ^{c)}	Conc. at sowing	Conc. at harvest	
Imidacloprid	4	0.733	0.740	0.662	0.332	0.330	0.412	
Clothianidin	4	0.918*	0.936**	0.868*	0.462	0.499	0.490	
Thiacloprid	4	0.975**	0.103	0.994**	0.123	0.170	0.062	
Metalaxyl	4	0.648	0.787*	0.670	0.844*	0.961**	0.852	
Fosthiazate	4	0.466	0.681	0.448	0.732	0.891*	0.750	
Flutolanil	4	0.566	0.691	0.481	0.543	0.511	0.511	
Procymidone	4	0.714	0.492	0.918*	0.489	0.551	0.359	
Tetraconazole	4	0.976**	0.985**	0.967**	0.334	0.328	0.289	

^{*} Significant at p<0.05; ** significant at p<0.01. a) Regression lines were calculated using data containing the point (0, 0), i.e., n=5. b) Soil-aging period (SAP) between application of pesticides and sowing. c) Mean values during the period between sowing and harvest.

the mean C_{total} of four test soils showed almost the same values, whereas the mean Cwater values of two andosols (soils PS3 and PS4), which had relatively high OC content, tended to be lower than those of non-andosoils (Supplemental. Figs. S4 and S5). Therefore, one reason for the high variability of residue concentrations in Komatsuna shoots for a given pesticide is considered to be the difference of the sorption strength, depending on the OC content of each test soil. However, we must consider that soil sorption is influenced not only by the OC content but also by the OC quality. For example, the sorption of PS4, which had the highest OC content, was lower than that of PS3, due to the difference in the chemical composition of the OC.¹¹⁾

To prevent the production and distribution of contaminated crops, pesticide concentrations in crops should be estimated before harvest with the appropriate technique. Table 3 shows a comparison of the R^2 values between pesticide concentrations in Komatsuna shoots and the mean Cwater values during the 28-day period until harvest and the C_{water} values at sowing or at harvest. The R² for many pesticides showed almost the same values, even when any C_{water} value was used. Hence, it is possible to estimate the residue concentrations in Komatsuna shoots by measuring the Cwater at sowing. However, thiacloprid and procymidone showed the lowest R^2 values when using the C_{water} values at sowing and the highest values when using the C_{water} values at harvest. Such results might be explained by the difference in the dissipation of the C_{water} of thiacloprid and procymidone during plant growth among test soils (Supplemental Tables S8–S11). The $\tau_{1/2}$ values in PS1, PS2, PS3, and PS4 were 4, 20, 40, and 11 days for thiacloprid, and 13, 52, 56, and 23 days for procymidone, respectively. The $au_{1/2}$ values differed depending on the soil type; $\tau_{1/2}$ was especially low in PS1 as compared with other soils. Among the test soils, the C_{water} was highest in PS1 at sowing, indicating that the absolute decrease of C_{water} in PS1 during plant growth was considerably higher than that in other soils. Thus, if a pesticide exhibits distinctly different dissipation of C_{water} among soils during plant growth, the estimation of the residue

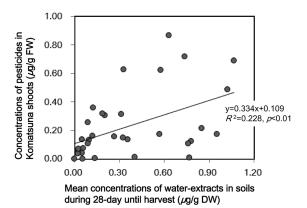


Fig. 2. Linear regression analysis between pesticide concentrations in Komatsuna shoots cultivated in four soils with a 7-day soil-aging period and the mean concentrations of water extracts in soils during a 28-day period until harvest, i.e., the period between sowing and harvest. The regression line was calculated using data containing the point (0, 0), *i.e.*, n=33.

concentrations in Komatsuna shoots using the C_{water} values at sowing is inadequate, and prediction of the C_{water} values at harvest is required. To predict the C_{water} values at harvest, it is necessary to evaluate the dissipation behavior of C_{water} . Many studies have investigated dissipation of Ctotal, including newly developed pesticides, whereas very few data for Cwater are available. The dissipation rate of C_{total} is different from C_{water} due to the longer $\tau_{1/2}$ of C_{total} as compared to that of C_{water} as a whole (Supplemental Tables S8-S11). Therefore, estimating pesticide residue in crops using predicted values of Cwater at harvest requires further study of the dissipation behavior of Cwater with respect to target pesticides and field soils in which crops are planted.

The applicability of the water-extraction method using C_{water} values at sowing differed with the type of pesticide; in some cases, to obtain more accurate levels of pesticides in Komatsuna shoots, the prediction of Cwater values at harvesting period was required. Therefore, the applicability of the method should be verified for each target pesticide based on the actual calibration

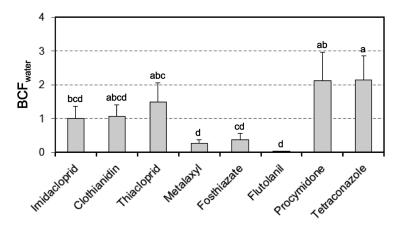


Fig. 3. Comparison of the bioconcentration factor (BCF_{water}), which was calculated by dividing the concentrations in Komatsuna shoots by the mean concentrations of water extracts in soil during a 28-day period until harvest, *i.e.*, the period between sowing and harvest. Columns with the same letter are not significantly different at p < 0.05, according to ANOVA-protected Tukey's multiple range testing. Error bars indicate standard deviations (n = 4).

curve.

On the other hand, when the data for all pesticides were plotted simultaneously, the correlation between the concentrations in Komatsuna shoots and the mean C_{water} values during the 28-day period until harvest was poor (Fig. 2). Such a weak correlation may be caused by the difference in the translocation property of pesticides to shoots. Hence, the bioconcentration factor (BCF_{water}) was calculated by dividing the concentrations in Komatsuna shoots by the mean C_{water} during the 28-day period until harvest (Fig. 3).

The mean BCF_{water} value for a given pesticide varied between 0.03 and 2.14; three pesticides-metalaxyl, fosthiazate, and flutolanil—showed low BCF_{water} values. In general, it has been postulated that translocation of organic chemicals to shoots is influenced by hydrophobicity. Briggs et al.23) found that the relationship between the transpiration stream concentration factor (TSCF), defined as the concentration of chemicals in the transpiration stream divided by the concentration in the external solution (i.e., hydroponic medium), and the $\log K_{ow}$ could be represented by a bell-shaped curve with a maximum TSCF at a $\log K_{\text{ow}}$ of 1.78, while Dettenmaier et al.²⁴⁾ reported a negative sigmoidal relationship between the TSCF and the $\log K_{ow}$. In the present study, however, the relationship between the BCF- $_{\text{water}}$ and the log K_{ow} was quite different from either of the above. The BCF_{water} values did not take the transpiration volume into account. However, the transpiration volume could be ignored when the pesticides were compared because the tested pesticides were simultaneously applied to a pot. Therefore, the translocation property of pesticides to shoots was not the direct cause of the difference in the BCF_{water}. On the other hand, the concentration of pesticides in plants is also affected by the plants' metabolism. For example, Fujisawa et al.25) improved the plant uptake models of nonionized pesticides using a metabolic parameter, i.e., the degradation rate constant in plants. Briggs et al.23) and Dettenmaier et al.²⁴⁾ studied the plant uptake of pesticides in hydroponic culture, and their experimental periods were 24-48 hr

and 5-50 hr, respectively. In addition, the TSCF values of Briggs et al.23) were corrected for degradation of chemicals in plant shoots. On the other hand, the present experiment was carried out using soil during the 28-day period and the BCF_{water} values were not corrected for plant metabolism. Therefore, there is a possibility that the experimental conditions, such as the exposure period, the presence or absence of soil, and correction for the degradation of chemicals in a plant, affect the difference between the BCF_{water} and the TSCF previously reported. However, there are no available data on the metabolism of tested pesticides in Komatsuna, which are taken up by the roots. The reasons for the variability in BCF water remain unexplained, warranting additional study. Therefore, at present, the water-extraction method for estimating the amount of pesticide residue in crops should not be applied across pesticides but only for a given pesticide.

Influence of soil-aging periods (SAPs) on pesticide concentrations in Komatsuna shoots

Fig. 4 shows the relationships between pesticide concentrations in Komatsuna shoots and three different SAPs-7-day, 35-day, and 63-day-for PS2 and PS3. The residue concentrations in Komatsuna shoots decreased with increasing SAPs. The ratios of the residue concentrations in Komatsuna shoots cultivated in soils with a 63-day SAP as compared to those cultivated in soils with a 7-day SAP were 29-90% (mean=50%) and 12-69% (mean=51%) in PS2 and PS3, respectively. Thus, the ratios differed according to the type of soil and the pesticide. Focusing on the effect of the aging period on the residue concentrations in soils, the ratios of the C_{total} values at sowing with a 63-day SAP as compared to those at sowing with a 7-day SAP were 27-108% (mean=80%) and 37-119% (mean=90%) in PS2 and PS3, respectively. On the other hand, the ratios of the C_{water} values at sowing with a 63-day SAP as compared to those at sowing with a 7-day SAP were 12-84% (mean=52%) and 17-69% (mean=43%) in PS2 and PS3, respectively. As mentioned above,

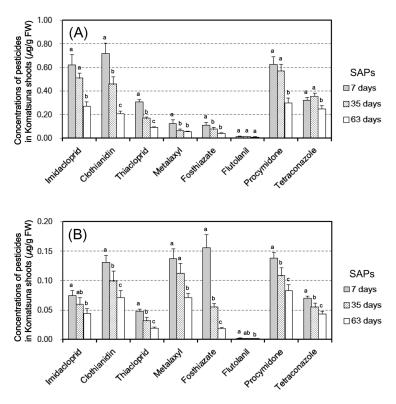


Fig. 4. Effect of soil-aging periods (SAPs) on residue concentrations of pesticides in Komatsuna shoots cultivated in soils PS2 (A) and PS3 (B). Columns with the same letter are not significantly different at p<0.05, according to ANOVA-protected Tukey's multiple range testing. Error bars indicate standard deviations (n=4).

Table 4. Linear regression analysis^{a)} between pesticide concentrations in Komatsuna shoots cultivated in soils with three different SAPs^{b)} and extractable concentrations in soils PS2 and PS3

Compound	No. soils	R^2							
		Water-extracts			Total-extracts				
		Mean conc. during 28-day until harvest ^{c)}	Conc. at sowing	Conc. at harvest	Mean conc. during 28-day until harvest ^{c)}	Conc. at sowing	Conc. at harvest		
Imidacloprid	2	0.922**	0.938**	0.882**	0.172	0.151	0.212		
Clothianidin	2	0.781**	0.802**	0.736*	0.227	0.150	0.285		
Thiacloprid	2	0.963**	0.908**	0.988**	0.003	0.045	0.004		
Metalaxyl	2	0.149	0.239	0.022	0.686*	0.585*	0.545		
Fosthiazate	2	0.339	0.476	0.204	0.787**	0.799**	0.756*		
Flutolanil	2	0.936**	0.807**	0.988**	0.306	0.293	0.260		
Procymidone	2	0.964**	0.868**	0.956**	0.207	0.270	0.120		
Tetraconazole	2	0.931**	0.726*	0.977**	0.379	0.371	0.343		

^{*}Significant at p<0.05; ** significant at p<0.01. a) Regression lines were calculated using data containing the point (0, 0), i.e., n=7. b) Soil-aging period (SAP) between application of pesticides and sowing. SAPs were set to 7 days, 35 days, and 63 days. c) Mean values during the period between sowing and harvest.

the dissipation of C_{total} was slower than that of C_{water} as a whole (Supplemental Tables S8–S11). Therefore, the reduction of C_{water} in soils caused by soil aging was larger than that of C_{total} . Several studies reported that soil sorption of pesticides such as imidacloprid, 26,27) nicosulfuron, 28) and simazine 29) increased during the aging period; thus, it is implied that the dissipation of C_{water} is affected not only by general dissipation factors such as biodegradation, hydrolysis, and volatilization, but also by increasing soil sorption with increased aging time.

To verify the effectiveness of pesticide reduction in Komatsuna shoots by setting long SAPs for a given residue concentration in soils, we elucidated the correlation between pesticide concentrations in Komatsuna shoots cultivated in soils with three different SAPs and the extractable residue in PS2 and PS3 (Table 4

and Supplemental Figs. S6 and S7). Similar to the four test soils with a 7-day SAP (Table 3), the R^2 values of the regression lines calculated using $C_{\rm water}$ were higher than those using $C_{\rm total}$, except for metalaxyl and fosthiazate. These results suggest that the water-extraction method of soil is more suitable for estimating pesticide concentrations in Komatsuna shoots cultivated in different soils with different SAPs than the total-extraction method. However, the R^2 values of the regression lines and the effectiveness of pesticide reduction in Komatsuna shoots by setting long SAPs changed depending on the combination of pesticides and soils. To apply the water-extraction method to target pesticides, it is necessary to prepare actual calibration curves, *i.e.*, regression lines, by using a target crop and field soils and to verify the applicability of the method based on the calibration curve.

As discussed above, the dissipation of C_{water} is considered to affect the pesticide concentrations in Komatsuna shoots. In the present study, dissipation studies of pesticides in soils were performed with pot experiments, in which the water content of soils was adjusted by bottom-up irrigation; thus, the dissipation of pesticides in the pots appeared slower than that in fields with leaching of pesticides after rainfall or irrigation.³⁰⁾ Therefore, the effectiveness of pesticide reduction in crops cultivated in pots by setting long SAPs may also be lower than that in fields. Hashimoto et al.4) reported that application of activated carbon to fields reduced dieldrin concentrations in cucumber fruits to 42-64% of those cultivated in non-treated fields. The effectiveness of dieldrin reduction in cucumber fruits by applying activated carbon roughly corresponds to the results obtained in the present study by setting long SAPs. Therefore, although the effectiveness of pesticide reduction differs depending on the type of pesticide and the soil, for pesticides that are available on the market, which show faster dissipation in soils than POPs, setting appropriate SAPs is suggested as one of practical measures to reduce pesticide concentrations in crops.

Conclusion

The water-extraction method of soils was more effective for estimating pesticide concentrations in crops cultivated in different soils with different SAPs than the total-extraction method. Furthermore, pesticide concentrations in Komatsuna shoots decreased by setting long SAPs. These results imply that estimating pesticide concentrations in crops based on the C_{water} and setting appropriate SAPs prevent production and distribution of crops contaminated with pesticide residues from soils. However, the applicability of the water-extraction method and the effectiveness of pesticide reduction in crops by setting long SAPs differed, depending on the combination of pesticides and soils. Therefore, a calibration curve for a target pesticide should be individually prepared by using the crop of interest and the field soil in which the crop is planted; the applicability of the waterextraction method should be verified from the results. Furthermore, the dissipation studies of C_{water} in actual fields are required to estimate the effectiveness of setting long SAPs for the reduction of pesticides in crops.

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